

Science: Long-Term Plan - Academic Year 2023 - 2024

Year 7/8	Autumn Term	Spring Term	Summer Term
	<u>Autumn 1</u>	<u>Spring 1</u>	<u>Summer 1</u>
	<p>Introduction to Science/key skills</p> <ul style="list-style-type: none"> ❖ use appropriate apparatus, during laboratory work, paying attention to health and safety ❖ ask questions and develop a line of enquiry based on observations ❖ make predictions using scientific knowledge and understanding ❖ select, plan and carry out the most appropriate types of scientific enquiries to test predictions, including identifying independent, dependent and control variables, where appropriate ❖ make and record observations and measurements ❖ present observations and data ❖ interpret observations and data, ❖ present reasoned explanations ❖ evaluate data ❖ understand and use SI units 	<p>Forces</p> <ul style="list-style-type: none"> ❖ forces as pushes or pulls, arising from the interaction between two objects ❖ using force arrows in diagrams, ❖ stretching and squashing - springs; ❖ friction between surfaces ❖ measuring of forces ❖ measurements of stretch or compression as force is changed ❖ contact and non-contact forces 	<p>Light</p> <ul style="list-style-type: none"> ❖ the similarities and differences between light waves and waves in matter ❖ light waves travelling through a vacuum ❖ speed of light ❖ the transmission of light through materials ❖ use of ray model to explain imaging in mirrors ❖ the pinhole camera ❖ the refraction of light ❖ the human eye ❖ colours and the different frequencies of light. <p>Sound</p> <ul style="list-style-type: none"> ❖ frequencies of sound wave ❖ measurement of sound waves ❖ echoes ❖ properties of sound ❖ how sound waves are produced ❖ the structure of the human ear ❖ auditory range of humans and animals

<u>Autumn 2</u>	<u>Spring 2</u>	<u>Summer 2</u>
<p>Basic Chemistry</p> <ul style="list-style-type: none"> ❖ atoms and molecules as particles ❖ The particulate nature of matter ❖ the properties of the different states of matter (solid, liquid and gas) ❖ Atoms, elements and compounds ❖ chemical symbols and formulae for elements and compounds ❖ the Periodic Table: periods and groups; ❖ metals and non-metals and their properties. <p>Cells & organisation</p> <ul style="list-style-type: none"> ❖ cells as the unit of living organisms ❖ how to observe, interpret and record cell structure using a light microscope ❖ the functions of the cell wall, cell membrane, cytoplasm, nucleus, vacuole, mitochondria and chloroplasts ❖ the similarities and differences between plant and animal cells ❖ the hierarchical organisation of multicellular organisms ❖ the role of diffusion in the movement of materials in and between cells ❖ the structural adaptations of some unicellular organisms 	<p>The skeletal and muscular systems</p> <ul style="list-style-type: none"> ❖ the structure and functions of the human skeleton ❖ biomechanics - the interaction between skeleton and muscles, ❖ measurement of force exerted by different muscles ❖ the function of muscles and examples of antagonistic muscles. <p>Gas exchange systems</p> <ul style="list-style-type: none"> ❖ the structure and functions of the gas exchange system in humans, including adaptations to function ❖ the mechanism of breathing to move air in and out of the lungs, using a pressure model to explain the movement of gases, including simple measurements of lung volume 	<p>Reactions</p> <ul style="list-style-type: none"> ❖ chemical reactions as the rearrangement of atoms ❖ representing chemical reactions using formulae and using equations ❖ combustion, thermal decomposition, oxidation and displacement reactions ❖ defining acids and alkalis in terms of neutralisation reactions ❖ the pH scale for measuring acidity/alkalinity; and indicators <p>Reproduction</p> <ul style="list-style-type: none"> ❖ reproduction in humans ❖ the structure and function of the male and female reproductive systems, ❖ menstrual cycle ❖ gametes, fertilisation, gestation and birth through the placenta ❖ reproduction in plants ❖ flower structure ❖ pollination, ❖ fertilisation ❖ seed and fruit formation ❖ seed dispersal

	Autumn Term	Spring Term	Summer Term
	<u>Autumn 1</u>	<u>Spring 1</u>	<u>Summer 1</u>
Year 9	Health & Lifestyle • the effects of recreational drugs (including substance misuse) on behaviour, health and life processes.	The Periodic Table • the varying physical and chemical properties of different elements • the principles underpinning the Mendeleev Periodic Table • the Periodic Table: periods and groups; metals and non-metals • how patterns in reactions can be predicted with reference to the Periodic Table • the properties of metals and non-metals • the chemical properties of metal and non-metal oxides with respect to acidity.	Adaptation & Inheritance • heredity as the process by which genetic information is transmitted from one generation to the next • a simple model of chromosomes, genes and DNA in heredity, including the part played by Watson, Crick, Wilkins and Franklin in the development of the DNA model • differences between species • the variation between individuals within a species being continuous or discontinuous, to include measurement and graphical representation of variation • the variation between species and between individuals of the same species means some organisms compete more successfully, which can drive natural selection • changes in the environment may leave individuals within a species, and some entire species, less well adapted to compete successfully and reproduce, which in turn may lead to extinction • the importance of maintaining biodiversity and the use of gene banks to preserve hereditary material.
	Elements, Compounds & Mixtures		
	Electricity	Magnetism • magnetic poles, attraction and repulsion • magnetic fields by plotting with compass, representation by field lines • Earth's magnetism, compass and navigation • the magnetic effect of a current, electromagnets, D.C. motors (principles only).	

	<u>Autumn 2</u>	<u>Spring 2</u>	<u>Summer 2</u>
	<p>Ecosystem processes</p> <ul style="list-style-type: none"> ♣ importance and process of photosynthesis ♣ the structure and function of a leaf including the distribution of chloroplasts in a leaf ♣ how plants use minerals for healthy growth ♣ the role of nitrates in plant growth ♣ the process of chemosynthesis and where it takes place ♣ importance and process of aerobic respiration ♣ the process and importance of anaerobic respiration ♣ the interdependence of organisms in an ecosystem, including food chains and webs ♣ how organisms affect, and are affected by, their environment, including the accumulation of toxic materials. ♣ resources that plants and animals compete for ♣ how organisms are adapted to their environments <p>Separation Techniques</p> <ul style="list-style-type: none"> ♣ the concept of a pure substance ♣ mixtures, including dissolving ♣ diffusion in terms of the particle model ♣ simple techniques for separating mixtures: filtration, evaporation, distillation and chromatography ♣ the identification of pure substances. <p>Forces & Motion</p> <ul style="list-style-type: none"> ♣ speed and the quantitative relationship between average speed, distance and time ($\text{speed} = \text{distance} \div \text{time}$) ♣ the representation of a journey on a distance-time graph ♣ relative motion: trains and cars passing one another 	<p>Chemical Reactions</p> <p>Energy</p> <ul style="list-style-type: none"> ♣ comparing energy values of different foods (from labels) (kJ) ♣ comparing power ratings of appliances in watts (W, kW) ♣ comparing amounts of energy transferred (J, kJ, kW hour) ♣ domestic fuel bills, fuel use and costs ♣ fuels and energy resources. <p>simple machines give bigger force but at the expense of smaller movement (and vice versa): product of force and displacement unchanged</p> <ul style="list-style-type: none"> ♣ heating and thermal equilibrium: temperature difference between two objects leading to energy transfer from the hotter to the cooler one, through contact (conduction) or radiation; such transfers tending to reduce the temperature difference: use of insulators ♣ other processes that involve energy transfer: changing motion, dropping an object, completing an electrical circuit, stretching a spring, metabolism of food, burning fuels ♣ energy as a quantity that can be quantified and calculated; the total energy has the same value before and after a change ♣ comparing the starting with the final conditions of a system and describing increases and decreases in the amounts of energy associated with movements, temperatures, changes in positions in a field, in elastic distortions and in chemical compositions ♣ using physical processes and mechanisms, rather than energy, to explain the intermediate steps that bring about such changes 	<p>Metals & Acids</p> <ul style="list-style-type: none"> ♣ compare the reactions of different metals with dilute acids, oxygen and water ♣ performing the test for hydrogen gas ♣ the reactivity series & metal displacement reactions ♣ extracting metals by heating with carbon ♣ properties and uses of ceramics ♣ properties and uses of polymers ♣ properties and uses of composites <p>Forces & Pressure</p> <ul style="list-style-type: none"> ♣ moment as the turning effect of a force ♣ pressure in gases, atmospheric pressure, decreases with increase of height as weight of air above decreases with height ♣ pressure liquids, increasing with depth; upthrust effects, floating and sinking ♣ pressure measured by ratio of force over area - acting normal to any surface.

Year 10	Autumn Term	Spring Term	Summer Term
	<u>Autumn 1</u>	<u>Spring 1</u>	<u>Summer 1</u>
	Atomic structure and radiation <ul style="list-style-type: none"> ❖ Atomic structure ❖ A simple model of the atom, symbols, relative atomic mass, electronic charge and isotopes ❖ Atoms and nuclear radiation ❖ The periodic table 	Bonding and structure (cont.) <ul style="list-style-type: none"> ❖ Chemical bonds, ionic, covalent and metallic ❖ How bonding and structure are related to the properties of substances ❖ Structure and bonding of carbon Energy <ul style="list-style-type: none"> ❖ Energy changes in a system, and the ways energy is stored before and after such changes ❖ Conservation and dissipation of energy ❖ National and global energy resources 	Non-communicable Disease Quantitative Chemistry <ul style="list-style-type: none"> ❖ Chemical measurements, conservation of mass and the quantitative interpretation of chemical equations ❖ Use of amount of substance in relation to masses of pure substances (HT) ❖ Concentration of solutions Chemical changes <ul style="list-style-type: none"> ❖ Reactivity of metals ❖ Reactions of acids ❖ Electrolysis
	<u>Autumn 2</u>	<u>Spring 2</u>	<u>Summer 2</u>
	Cell Biology <ul style="list-style-type: none"> ❖ Cell structure ❖ Cell division ❖ Transport in cells Particle model of matter <ul style="list-style-type: none"> ❖ Changes of state and the particle model ❖ Internal energy and energy transfers ❖ Particle model and pressure Bonding and structure. <ul style="list-style-type: none"> ❖ Chemical bonds, ionic, covalent and metallic ❖ How bonding and structure are related to the properties of substances ❖ Structure and bonding of carbon 	Energy changes <ul style="list-style-type: none"> ❖ Exothermic and endothermic reactions Organisation: Animals <ul style="list-style-type: none"> ❖ Principles of organisation ❖ Animal tissues, organs and organ systems Bioenergetics: Animals <ul style="list-style-type: none"> ❖ Aerobic and anaerobic respiration ❖ Response to exercise ❖ Metabolism Communicable Disease <ul style="list-style-type: none"> ❖ Communicable (infectious) diseases ❖ Human defence systems ❖ Vaccination ❖ Antibiotics and painkillers ❖ Discovery and development of drugs 	Organisation: Plants <ul style="list-style-type: none"> ❖ Plant tissues, organs and systems Bioenergetics: Plants <ul style="list-style-type: none"> ❖ Photosynthetic reaction ❖ Rate of photosynthesis ❖ Uses of glucose from photosynthesis Electricity <ul style="list-style-type: none"> ❖ Current, potential difference and resistance ❖ Series and parallel circuits ❖ Domestic uses and safety ❖ Energy transfers

	Autumn Term	Spring Term	Summer Term
Year 11	<u>Autumn 1</u>	<u>Spring 1</u>	<u>Summer 1</u>
	Year 10 Biology Catch up <ul style="list-style-type: none"> ❖ Non-communicable & Communicable Disease ❖ Organisation: Plants ❖ Bioenergetics: Plants Year 10 Chemistry Catch up <ul style="list-style-type: none"> ❖ Energy changes & Chemical changes ❖ Quantitative Chemistry Year 10 Physics Catch up <ul style="list-style-type: none"> ❖ Radiation & Electricity 	Using resources <ul style="list-style-type: none"> ❖ Using the Earth's resources and obtaining potable water ❖ Life cycle assessment and recycling Rate of reaction <ul style="list-style-type: none"> ❖ Rate of reaction ❖ Reversible reactions and dynamic equilibrium 	Inheritance, variation and evolution <ul style="list-style-type: none"> ❖ Reproduction ❖ Variation and evolution ❖ The development of understanding of genetics and evolution ❖ Classification of living organisms Magnetism and electromagnetism <ul style="list-style-type: none"> ❖ Permanent and induced magnetism, magnetic forces and fields & The motor effect
	<u>Autumn 2</u>	<u>Spring 2</u>	<u>Summer 2</u>
	Homeostasis and response <ul style="list-style-type: none"> ❖ Homeostasis ❖ The human nervous & endocrine system ❖ Control of blood glucose concentration ❖ Hormones in human reproduction, Contraception ❖ The use of hormones to treat infertility (HT) ❖ Feedback systems (HT) Forces <ul style="list-style-type: none"> ❖ Forces and their interactions ❖ Work done and energy transfer ❖ Forces and elasticity & Forces and motion ❖ Accelerations and Newton's Laws of motion ❖ Forces and braking ❖ Momentum (HT only) Chemistry of the atmosphere <ul style="list-style-type: none"> ❖ The composition and evolution of the Earth's atmosphere ❖ Carbon dioxide and methane as greenhouse gases ❖ Common atmospheric pollutants and their sources 	Organic chemistry <ul style="list-style-type: none"> ❖ Carbon compounds as fuels and feedstock Chemical analysis <ul style="list-style-type: none"> ❖ Purity, formulations and chromatography ❖ Identification of common gases Waves <ul style="list-style-type: none"> ❖ Waves in air, fluids and solids ❖ Electromagnetic waves 	Ecology <ul style="list-style-type: none"> ❖ Adaptations, interdependence and competition ❖ Organisation of an ecosystem ❖ Biodiversity and the effect of human interaction on ecosystems <p>Revision</p>